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Test if you are looking for a way to test out just how much you know about them. For you to pass this quiz, you will need to have some knowledge when it comes to formulas of different acid salts and base and their properties. Do give it a shot, and feel free to take the quiz as much as you need.

**Quiz: Take This
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Page 5/23

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Which of the following values would represent the pH of a strong base?

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Bases react with a.
acids to produce salts
and water. c. water to

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produce acids and
salts. b. salts to
produce acids and
water. d. neither acids,
salts, nor water. ... Acid
Base Practice Test
Answer Section

MULTIPLE CHOICE 1.

ANS: B DIF: I OBJ:

15-1.1 2. ANS: C DIF: I

OBJ: 15-1.1 3. ANS: A

...

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Acids Bases and Salts
important question for

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acids mixed with bases
make stronger acids

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Acids and bases interact with each other in what is called a neutralization reaction. The products of the reaction are a salt and water. The pH is "neutralized" to 7 if both the acid and base fully react.

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The following table shows the neutralisation equations when acid reacts with metal, base, metal oxide, metal carbonate,

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ammonia, and when
base reacts with
ammonium salt, non-
metal oxide. Scroll
down the page for
examples and
explanations. A salt is
an ionic compound that
can be formed by the
neutralization reaction
of an acid and a base.

Acid, Bases, Salts - IGCSE Chemistry (solutions, examples

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Answer: Basic solutions turn red litmus paper blue. The salt of a weak acid and a strong base gives a basic solution. SO, the given salt X is the salt of a weak acid and a strong base. . Example : When sodium carbonate is dissolved in water, it gets hydrolysed to some extent and forms sodium hydroxide and carbonic acid.

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Salts Class 10 Extra Questions with ...

In acid - base chemistry, salts are ionic compounds that result from the neutralization reaction of an acid and a base. Basic salts contain the conjugate base of a weak acid, so when they dissolve in water, they react with water to yield a solution with pH greater than 7.0.

Key Terms. basic salt: the product of the

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neutralization of a strong base and a weak acid; its anion is the conjugate base of the weak acid

Acid-Base Properties of Salts | Boundless Chemistry

Some of these substances taste sour, some taste bitter, some taste sweet and some taste salty.

Acids, bases and salts are the three important groups of chemical

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substances that are used by us in different ways. Some of the acids, bases and salts occur in nature and they can be made artificially in factories also. Edible substances and their tastes

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