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Econometrics Final Exam And Solutions

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To convert these responses to a pH reading, the electrode must be calibrated against standard solutions of known H concentration. Determine the weight in grams of sodium dihydrogen phosphate ($\text{NaH}_2\text{PO}_4 \cdot \text{H}_2\text{O}$; FW 138) and disodium hydrogen phosphate (Na_2HPO_4 ; FW 142) needed to prepare 1 L of a standard buffer at pH 7 with a total phosphate concentration of 0 M (see Fig. 2-15). See ...

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